## Syllabus for General Chemistry Laboratory I

#### 2025 SPRING Semester

## Course Information

1) Course Number: CH102

2) Lecture: Experiment: credit = 0:3 h:1

3) Location: Lab 402, 406 & 704 in Goong-Ni Lab Building (E6-5)

		Mon		Tue		Wed		Thu		Fri	
		Class	А	Class	G	Class	М	Class	S	Class	
		Prof	김상규	Prof	김상규	Prof	김상규	Prof	김상규	Prof	
a.m.	9:00 ~	Class	В	Class	Н	Class	N	Class	T	Class	Y
a.iii.	12:00	Prof	최인성	Prof	최인성	Prof	최인성	Prof	최인성	Prof	김상규
		Class	С	Class	I	Class	0	Class	U	Class	Z
		Prof	김현우	Prof	김현우	Prof	김현우	Prof	김현우	Prof	최인성
		Class	D	Class	J	Class	Р	Class	V	Class	AA
		Prof	처칠	Prof	처칠	Prof	처칠	Prof	처칠	Prof	김현우
p.m.	13:00 ~	Class	Е	Class	K	Class	Q	Class	W	Class	AB
φ	16:00	Prof	전용웅	Prof	전용웅	Prof	전용웅	Prof	전용웅	Prof	처칠
		Class	F	Class	L	Class	R	Class	Х	Class	AC
		Prof	박윤수	Prof	백윤정	Prof	박윤수	Prof	백윤정	Prof	전용웅

# <u> Laboratory Manual (Textbook)</u>

1) Major text book: Laboratory manual for principles of general chemistry 10/e, Beran

\*Supplementary Resource: JOVE resources (<a href="https://app.jove.com/education/chemistry-list">https://app.jove.com/education/chemistry-list</a>)

## Objectives for the Course

- 1) To teach basic laboratory techniques.
- 2) To introduce elementary methods of assessing the significance of experimental measurements.
- 3) To provide an experience that enables the students to acquire a positive attitude toward chemistry, or science.

### Course Requirements

- 1) Preparation in advance for experimental work.
  - (a) Pre-laboratory reading preparation for successful experimentation.
  - (b) Pre-lab quizzes are, or pre-lab reports are required relating to the experimental details.
- 2) Writing up and submission of lab reports. Includes the following (but see evaluation):
  - (a) Observations and experimental detail.
  - (b) Detailed method of processing the experimental data.
  - (c) (For Quantitative analytical experiments)

Calculations and Conclusions regarding the accuracy and the precision of experimental results and errors and the inherent errors based on the measurements.

## **Grading**

Less than 50% of the total students in each class receive an A grade. Students who score below 50% of the total marks or fail to attend two or more experiments are assigned an F grade.

### **Schedule**

- Exp.o. Introduction, Lab Schedule, and Method of Evaluation
- Exp.1. Atomic and Molecular Structure (Experiment Dry Lab3)
- Exp.2. Periodic Table and Periodic Law (Experiment 11)
- Exp.3. Quantum Chemical Calculation: The Potential Energy Curve and the Orbitals of H<sub>2</sub><sup>+</sup>
- Exp.4. A Carbonate Analysis: Molar Volume of Carbon dioxide (Experiment 13)
- Exp.5. Thermodynamics of the Dissolution of Borax (Experiment 26)
- Exp.6. A Rate Law and Activation Energy (Experiment 24)
- Exp.7. LeChatelier's Principle; Buffers (Experiment 16)
- Exp.8. Alkalinity of a Water Resources (Experiment 20)
- Exp.9. Galvanic Cells, the Nernst Equation (Experiment 32)

Period		E	xperiment #	Notes			
renod	Mon	Tue	Wed	Thu	Fri	(Exp#/Lab#/Class)	
2/24 ~ 2/28	[No class]	[No class]	[No class]	[No class]	[No class]		
3/3 ~ 3/7		[Expo]	[Expo]	[Expo]	[Expo]	Expo/402/G1, Expo/406/G2, Expo/704/G3 3/3 (Samiljeol-substitute holiday)	
3/10 ~ 3/14	[Expo]	[Exp1~3]	[Exp1~3]	[Exp1~3]	[Exp1~3]	Exp1/402/G1, Exp2/406/G2, Exp3/704/G3	
3/17 ~ 3/21	[Exp1~3]	[Exp1~3]	[Exp1~3]	[Exp1~3]	[Exp1~3]	Exp1/402/G3, Exp2/406/G1, Exp3/704/G2	
3/24 ~ 3/28	[Exp1~3]	[Exp1~3]	[Exp1~3]	[Exp1~3]	[Exp1~3]	Exp1/402/G2, Exp2/406/G3, Exp3/704/G1	
3/31~ 4/4	[Exp1~3]	[No class]	[Exp4~6]	[Exp4~6]	[Exp4~6]	Exp4/402/G1, Exp5/406/G2, Exp6/704/G3  (Monday classes: Exp1/402/G1,  Exp2/406/G2, Exp3/704/G3)	
4/7 ~ 4/11	[Exp4~6]	[Exp4~6]				Exp4/402/G1, Exp5/406/G2, Exp6/704/G3	
4/14 ~ 4/18							
4/21 ~ 4/25	[Exp4~6]	[Exp4~6]	[Exp4~6]	[Exp4~6]	[Exp4~6]	Exp4/402/G3, Exp5/406/G1, Exp6/704/G2	
4/28 ~ 5/2	[Exp4~6]	[Exp4~6]	[Exp4~6]	[Exp4~6]	[Exp4~6]	Exp4/402/G2, Exp5/406/G3, Exp6/704/G1	
5/5 ~5/6~ 5/9						5/5:Children's Day, 5/6 (Buddha's Birthday -substitute holiday)	
5/12 ~ 5/16	[Exp7~9]	[Exp7~9]	[Exp7~9]	[Exp7~9]	[Exp7~9]	Exp7/402/G1, Exp8/406/G2, Exp9/704/G3	
5/19 ~ 5/23	[Exp7~9]	[Exp7~9]	[Exp7~9]	[Exp7~9]	[Exp7~9]	Exp7/402/G3, Exp8/406/G1, Exp9/704/G2	
5/26 ~ 5/30	[Exp7~9]	[Exp7~9]	[Exp7~9]	[Exp7~9]	[Exp7~9]	Exp7/402/G2, Exp8/406/G3, Exp9/704/G1	
6/2 ~ <mark>6/6</mark>						6/6: Memorial Day	
6/9 ~ 6/13							

Course Web Site for General Chemistry and General Chemistry Laboratory: <a href="http://www.gencheminkaist.pe.kr/">http://www.gencheminkaist.pe.kr/</a>

G2 classes: B, E, H, K, N, Q, T, W, Y, AB/ G3 classes: C, F, I, L, O, R, U, X, Z, AC)

### Methods of Evaluation

The student will be evaluated based on his/her performance of the requirement listed below.

#### One Experiment = 100 points

### (9 Experiments X 100 points) + (Safety Training: 100 points) = 1000 points in total

1. Laboratory experiment

The assignment description of an experiment consists of three parts; Pre-lab Assignment, Attitude, and Lab reports. Point distribution is as follows.

- 1) Pre-Lab Assignment (20 points)
  - (a) Introduction (Theory) and Procedure Summary
  - (b) Quiz
- 2) Attitude (20 pts)
  - (a) Lateness, lab safety, or cleanup
  - (b) Concentration or comprehension
- 3) Lab Reports (60 points)
  - (a) Result
  - (b) Discussion
    - Summary
    - Assessing the results
    - Conclusions
  - (c) Reference
  - (d) Post-lab questions
- 2. Online Safety Training Completion (Period: March 17 ~ April 16)
  - 1) Route: safety.kaist.ac.kr -> Lab Safety Management System-> Lab Safety Education -> Online Education
  - 2) Bachelor's Requirement

Classification	Hours Required	Topics
Freshmen	<ul> <li>New user education for freshmen-2 hours (within 3 months of admission) +</li> <li>Regular Education: 6 hours</li> </ul>	<ul> <li>New user education: [New Education]</li> <li>Regular education: Free choice of</li> <li>Chemistry Topics</li> </ul>
Sophomore ~ Senior ~	Regular education: 6 hours ~	Regular education: Free choice of Chemistry  Topics

<sup>\*</sup>KS Lab (A laboratory subject to thorough safety inspection): General Chemistry Lab (E6-5)

Lists	Section	Penalty /Bonus Points	Explanations				
Pre-lab	summary (5 pts each)		Introduction and procedure s	ummary on the lab manual			
assignment (20 pts)	Quiz (10 pts)		Laboratory Question(s) or Lab manual				
	Attendance: late coming		When you arrive at the lab before TA calls your name, you get a full score, 5 pts.  If you arrive 10 min late after the beginning of the lab, you get 3 pts. When you arrive between 10 min and 20 min, you will get the score, 0 pts. For being 20 minutes late, you can't participate in the lab experiment. (No chance anymore.)				
	Attendance: absence without o notice		If you are absent 3 times without any notices, F credit will be recorded.  *The lab experiment is ONLY permitted in registered class.				
			*Only valid excuses are illness (with a valid doctor's excuse) or unusual circumstance beyond your control (death in family), etc.  * The student must submit written verification of them for an excused absence.				
Attitude (20 points)	Lab safety & Cleanup	-5	Personal Protective Equipment: Safety glasses, Attire (lab coat) Wear approved eye protection while in the laboratory. The penalty is -5 pts for this. Care of balance: Anyone found to be leaving spilled chemicals in the balance area or bench will lose 5 pts for each violation. Any chemicals spilled on the benchtop or the balance pan during transfer are to be cleaned up immediately by the student. Cleanup of Lab Bench: The students are responsible for cleaning up their immediate lab bench area before leaving lab. This means wiping up any spills and disposing of any paper towels from the benchtop and sink. Failure to do this result in a deduction of -5 pts. Sink Disposal of chemicals: DO NOT discard any chemicals down the laboratory sink. The penalty is -5 pts for this.				
	Concentration &	-5	Check list - playing with gossiping - cell phone (touching, ringing, chatting), and so on - Repeated failure of the same experiment				
			o Report Format				
			Date				
			Name Co-worker				
Laboratory	Result &		MSDS for chemicals	Summary Section2: Hazards identification Section 4: First aid measures Section 5: Firefighting measures Section 7: handling and storage Section 8: exposure controls/personal protection Section 10: stability and reactivity Section 11: toxicological information			
Report (6o points)	Discussion (50 pts)		Results	o Data or Analysis o Calculations (with units) o Graphs o Tables			
			Discussion (20 ~ 40 lines)	o Summary o Assessing the results (analysis) o Conclusions			
			Reference				
			*Lab partner will share data; *Duplicate reports will be give	t Summary to TA before leaving lab. however separate lab report must be written by each student. en a zero. t or resubmit your work as theirs will result in a zero for both			
	Post-lab questions (10pts)		You should solve the question *Identical question answers v	ns and submit them in lab class to your TA. vill receive a zero.			

## **Laboratory Makeup Policy**

Make-ups are only for those who missed labs for a legitimate reason and got TA's permission to make up lab. (Give a written documentation to your lab TA during the next lab or bring it to his lab and ask permission to make up the missed lab)
Missed labs without a valid excuse, shall count as zero (o) and the student will not be allowed to make up this missed lab.
Students are not allowed to make up late lab reports or to make up more than two experiments during the scheduled lab make-up time. Re-doing labs is not possible.
Missed scheduled make-ups will be counted as zero and will not be rescheduled unless exceptional circumstances exist.
Your lab TA will let you know the make-up schedule and the due dates for make-up reports.

## Assignment Submission and Feedback

- Online Submission: <u>www.turnitin.com</u>
  - Create your account. Enter Class ID and Password (provided by TA).
  - Enter your name in Korean or English.
- Submission Due: Within 7 days
- Posting Grade: Within 3 days from the due date
- Claim Period: Within 2 days after the period of grading and feedback

### [Example]

Wed	Thu	Fri	Sat	Sun	Mon	Tue
	[Exp1]			Due: Within 7 da	ays	
	Period for G	rading and Feed Assignment	dback of the	Claim	n period	

### Policy for Late Lab Report

You should submit your assignment by due date on Turnitin. You are entirely responsible for both *upload of* the assignment file within due date and confirmation of the successful upload to Turnitin. Warning! We will not accept any excuses or compromise in case that you deduct any points below due to your late report submission. If you miss last chance, you don't need to submit it.

- ✓ Within 24 hours (last chance): -20 points
- ✓ More than 24 hours: -100 points

### Our Guidelines to check Plagiarism Using the Turnintin Software

Reports having similarity ranges from 24% to 100% or the following common sources will be regarded as plagiarism that results in zero for all reports involved (determined by chief TA and instructor).

#### Example 1. Text matching

- Overall Similarity index: Should not exceed 24% (24% and below gives the color code-Green, in Turnitin, indicating that it within the acceptable).
- Single source similarity index: Should not exceed 2%
- Acceptable number of words in unbroken string (phrase or sentence): Should not exceed 15 words

The color of the report icon indicates the similarity score of the report, based on the amount of matching or similar text that was uncovered. The percentage range is 0% to 100%. The possible similarity ranges are:

- Blue: No matching text
- Green: One word to 24% matching text
- Yellow: 25-49% matching text
- Orange: 50-74% matching text
- Red: 75-100% matching text

TITLE	SIMILARITY
Submission	0%
Submission	6%
Submission	43%
Submission	58%
Submission	80%

If you get an orange 52% similarity percentage, that means that 52% of your paper is exactly the same as other sources found by Turnitin. Even in case of 15% similarity, if the matching text is one continuous block of borrowed material (should not exceed 15 words), it will be considered as plagiarized text of significant concern.

**Example 2**. Cut/ Copy and Paste material from the Web, textbooks or online manual, data (lifting phrases, sentence and paragraphs of someone's work beyond an acceptable number of words)

Example 3. Copying the work of another student